

CO₂ Capture by Amine Scrubbing

By

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Roadmap

- Research on Amine Scrubbing
 - Objectives and Scope
 - Goals
- Conclusions on Energy
 - Amine Scrubbing is THE Process
- Conclusions on Solvent Management
 - Amine Scrubbing is THE Process
- R&D needs for amine scrubbing

Objectives of the Luminant Program

- Train graduate & undergraduate students
- Create understanding, data, & methods to facilitate deployment of CO₂ capture
- Create innovations to enhance performance and reduce cost of CO₂ capture

- Strategy

- Apply ChE science to understand & quantify the performance of MEA absorption/stripping
 - Energy Performance
 - Solvent Management
- Develop evolutionary innovations
 - Better Processes
 - Better Solvents

Annual Funding by Important Sponsors

Current

- \$500k Luminant Energy
- \$800k 29 companies
- \$800k State of Texas – ovhd & faculty
- \$2100k Total Annual Program Funding

Pending

- \$350k NRG/DOE, pilot & tech support
- \$325k CO₂ capture pilot plant project (C2P3)
- \$1000k DOE (w URS & Trimeric), pilot testing

Energy Science

- Thermo and Rates
 - FTIR: P_{CO_2} , P_{amine} : 30-70C, 120-160C
 - P_t : 120-160C
 - Wetted Wall Column: P_{CO_2} , k'_g : 40-100C
 - C_p by DSC, Speciation by NMR
- Packing Characterization in 18-inch column
 - a (CO_2), $K_g a$ (SO_2), $k_l a$ (toluene)
- AspenPlus Process Modeling
 - Thermo by Electrolyte NRTL
 - Diffusion & kinetics by RateSep
 - Dynamics by ACM
- Validation by 18-inch absorber/stripper
 - Air/ CO_2

Science of Solvent Management

- Oxidative Degradation
 - With FTIR – 1-5 days – 40-70C
 - 1-4 weeks, low gas flow, 40-70C
 - Cyclic: 2 weeks, 70-130C
- Thermal degradation - cylinders at 100-175C
- Nitrosoamine production & analysis
- Amine Volatility by FTIR
- Reclaiming: Thermal, Sulfate Crystallization
- Analytical tools
 - HPLC: cation, anion, aminoacid, molecule
 - MS with HPLC, cation, GC

Process Innovation Goals

- Use Heat rather than mechanical compression
 - Strip at 120-150°C, limited by degradation
- Reversible Stripping – high lean Idg, 0.3-0.4
- Maximize Heat Reversibility at stripper
 - Cross Exchanger & Reboiler $\Delta T=5^{\circ}\text{C}$
 - 2-stage flash, matrix, multiP, or interheat
 - Vapor recompression
- Minimize T of Absorption, 30-40°C
 - Precooling & intercooling Gas & Solvent
 - Control water balance
- Simplify

Solvent Innovation Goals

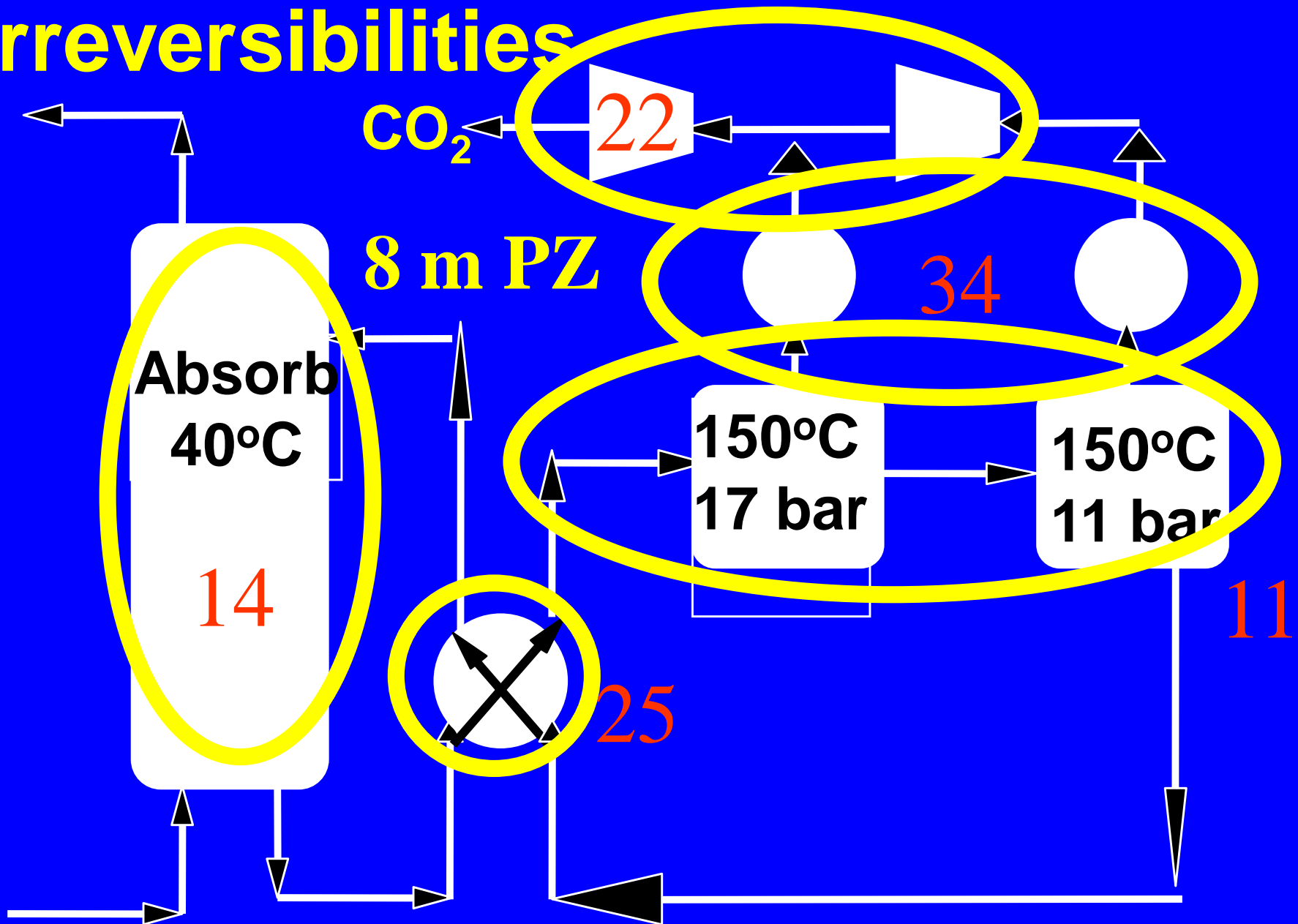
- Cost <\$5/lb, one step past MEA
- CO₂ Rate >MEA
- Heat of Absorption >70 kJ/mole
- T Stability >140°C
- Reversible Capacity >0.8 mole/kg
- Viscosity <15 cP
- No solid or liquid phase separation
- Low oxidative degradation
- Volatility=10-30 ppm, allows reclaiming

Solvents

- Intense Work
 - 7 to 11 m (30-40%) MEA
 - 8 m (40%) Piperazine (PZ)
 - 7 m MDEA/2 m PZ (50 wt)
 - Piperazine Derivatives
- Screening
 - Aminoacids
 - Hindered amines
 - DGA, EDA, MAPA

Conclusions on Energy

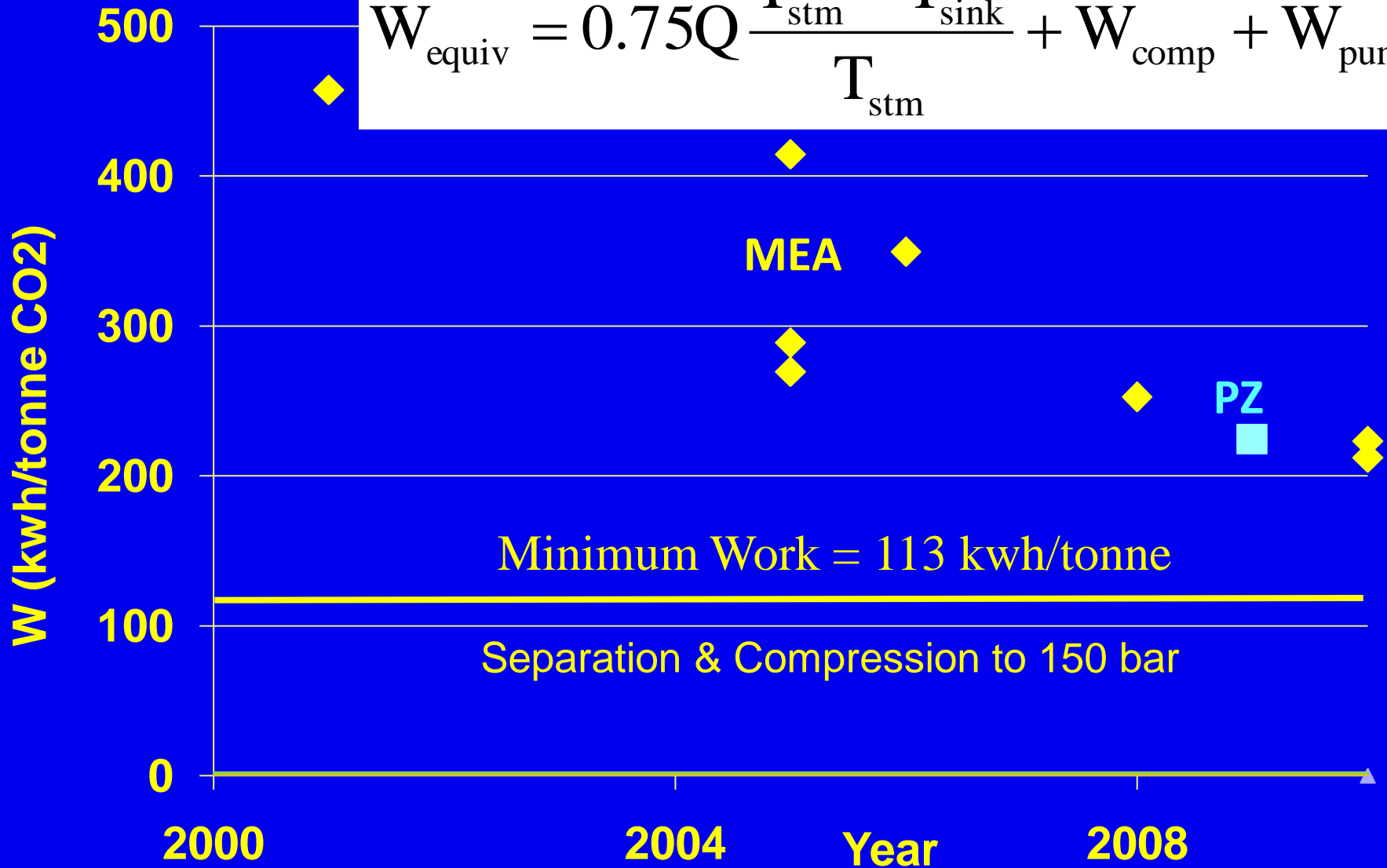
Irreversibilities



$W_{\text{ideal}} = 113 \text{ kwh/tonne}, W_{\text{real}} = 219 \text{ kwh/tonne}$

Equivalent Work of Amine Scrubbing

$$W_{\text{equiv}} = 0.75Q \frac{T_{\text{stm}} - T_{\text{sink}}}{T_{\text{stm}}} + W_{\text{comp}} + W_{\text{pump}}$$



Separation Efficiency, W_{\min}/W (%)

CO ₂ capture by 8 m PZ (2-stage regeneration at 150°C)	50
Cryogenic air separation (Darde et al., 2008)	25
Hydrocarbon Distillation (relative volatility=1.1 to 5)	10-36
Desalination by reverse osmosis (Semiat, 2008)	21

Capture processes driven by mechanical compression are not competitive

	Work (kwh/tonne)	Efficiency (%)
Minimum Work (Isothermal, ideal compression)	111	100
PZ with 2-stage regeneration at 150C	219	50
Ideal process driven by real compression 72% eff, 40°C cooling, $P_j/P_{j-1} = 2$ (Ideal Membranes or PSA)	206+	55
Oxycombustion with Ideal air separation Real compressors for Air & CO ₂ 1.35 moles O ₂ /mole CO ₂	217+	52
Real Oxycombustion (Darde et al., 2008)	284	40

Thermal Swing Regeneration of solids requires difficult equipment.

- Low absorption T with high heat of absorption
- Solids regeneration at high pressure
- “Cross exchange” at $\Delta T=5^{\circ}\text{C}$
- And low cost, robust solids

Conclusions

Research on Solvent Management

MEA is inexpensive & stable to 120C but degrades readily with O₂ or high T

PZ is more expensive but stable to 150C & oxidizes slowly

Both are reclaimed by evaporation

High cost reagents or materials may not survive coal flue gas

- Continuous & intermittent impurities
 - SO_3 , SO_2 , HCl, fly ash, NO_x , O_2 , H_2O
 - Fe^{+2} , Cr^{+3} , gypsum, CaCO_3 , T excursions
- High cost materials must be reclaimable
- High cost materials will require costly pretreating
- Most likely a fatal flaw for anything $>\$5/\text{lb}$
 - Designer amines
 - Enzymes
 - Ionic liquids
 - Membranes
 - Most organic & inorganic adsorbents/absorbents
 - Nanomaterials

Some R&D Needs for Amine Scrubbing

- Solvent Management
 - Nitrosamines, aldehydes, and other toxics
 - Oxidative Degradation
 - Reclaiming
 - Pilot plant testing with coal flue gas
- Energy
 - Inexpensive solvents with greater heat of absorption
 - Inexpensive solvents stable to higher T
- Scale-up
 - Advanced strippers
 - Absorbers
 - Exchangers: Reboilers, Exchangers
 - Steam Extraction
 - Pretreating & water balance

History Repeats

CaCO₃ Slurry:::Amine Scrubbing

CaCO ₃	Event	Amine
1948	1 st commercial plant	1980
1970	Too commercial for Gov. support But too costly, too dirty to use	1990
1970-82	Government funds advanced alts	1995-
1975-85	Govern. & EPRI fund test facilities	2010-
1977	Power Industry deploys 500+ MW	2016?
2009	First choice dominates	2030 ?

A Newer System to Reduce cost by 10-20%

8 m (40 wt%) Piperazine

- 10-20% less energy than 30 wt% MEA
 - 2 x CO₂ mass transfer rate
 - 1.8 x capacity
 - High P (6 – 15 atm) Stripper, stable to 150°C
- Oxidatively stable, esp. with Inhibitor A
- Less volatile than 7 m MEA
- Good Opportunities for Reclaiming

Economic Perspective

- Capture is \$50-75/tonne CO₂ (\$50-75/MWH)
 - Wholesale power doubles (\$50 to 100/MWH)
- Breakeven w Gas at \$70/MWH (capture \$50)
 - Existing plants, fuel + capture
 - Coal @\$2/MMBtu
 - Gas @ \$4.50/MMBtu
- Gasoline @ \$0.40/gal = \$50/Tonne CO₂

Aqueous Abs/Str: Near commercial

- 100's of plants for treating H₂ & natural gas
 - MEA and other amine solvents
- 10's of plants with natural gas combustion
 - Fluor, 30% MEA
 - MHI, KS-1
- A few plants with coal combustion
 - Abb-Lummus, 20% MEA, 6, 8, & 33 MW
 - Fluor, 30% MEA, (6 MW)
 - MHI, KS-1 (7 & 25 MW)
- Still need coal on 800 MW

One Conclusion on Energy

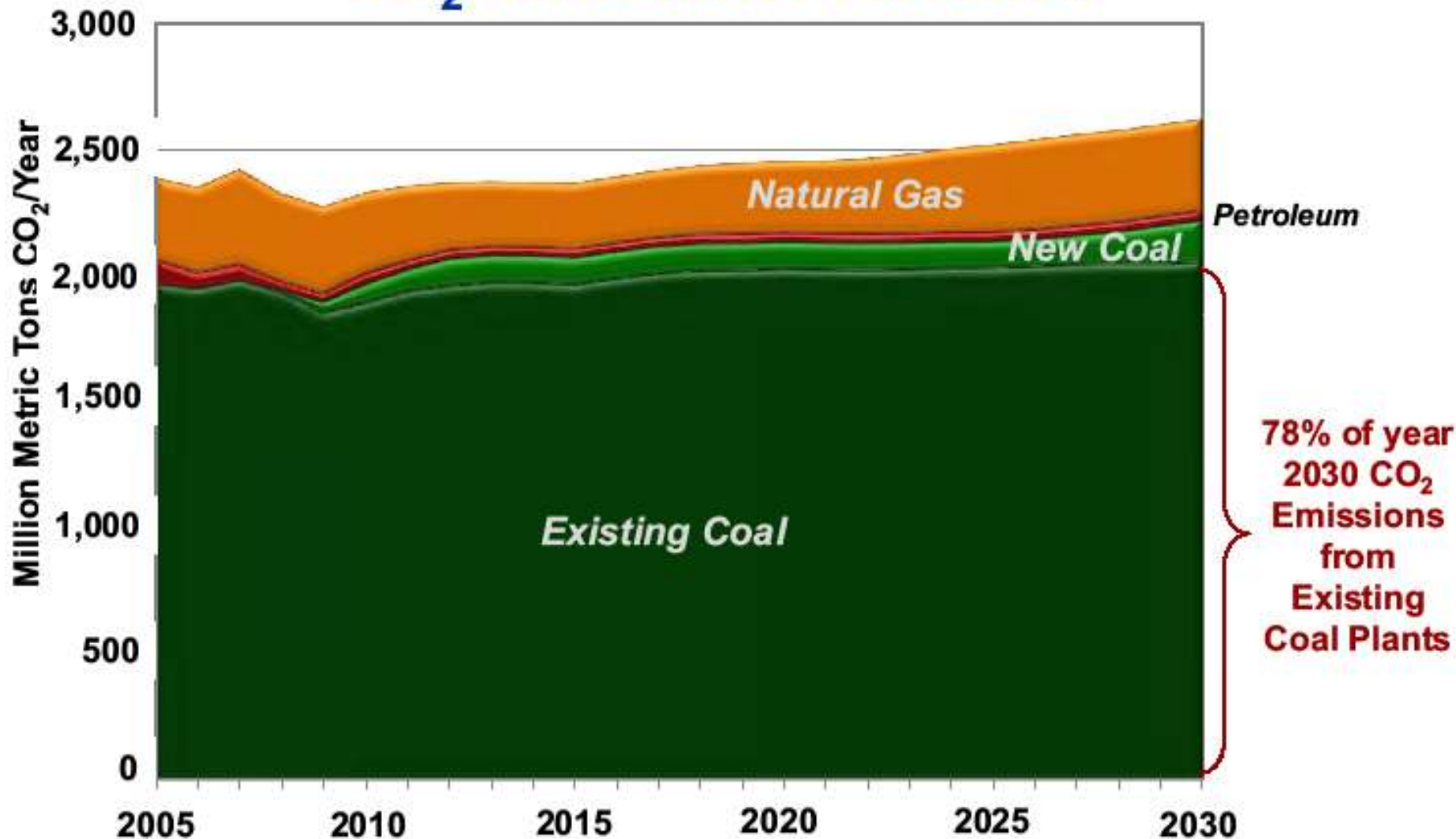
8 m PZ (40 wt%) with 2-stage Flash at 150°C

- 10-20% less energy than 30 wt% MEA
 - 2 x CO₂ mass transfer rate
 - 1.8 x capacity
 - Stable to 150°C
 - Stripping at 11 & 17 atm
- $W_{\min}/W_{\text{actual}} = 50\%$ efficiency

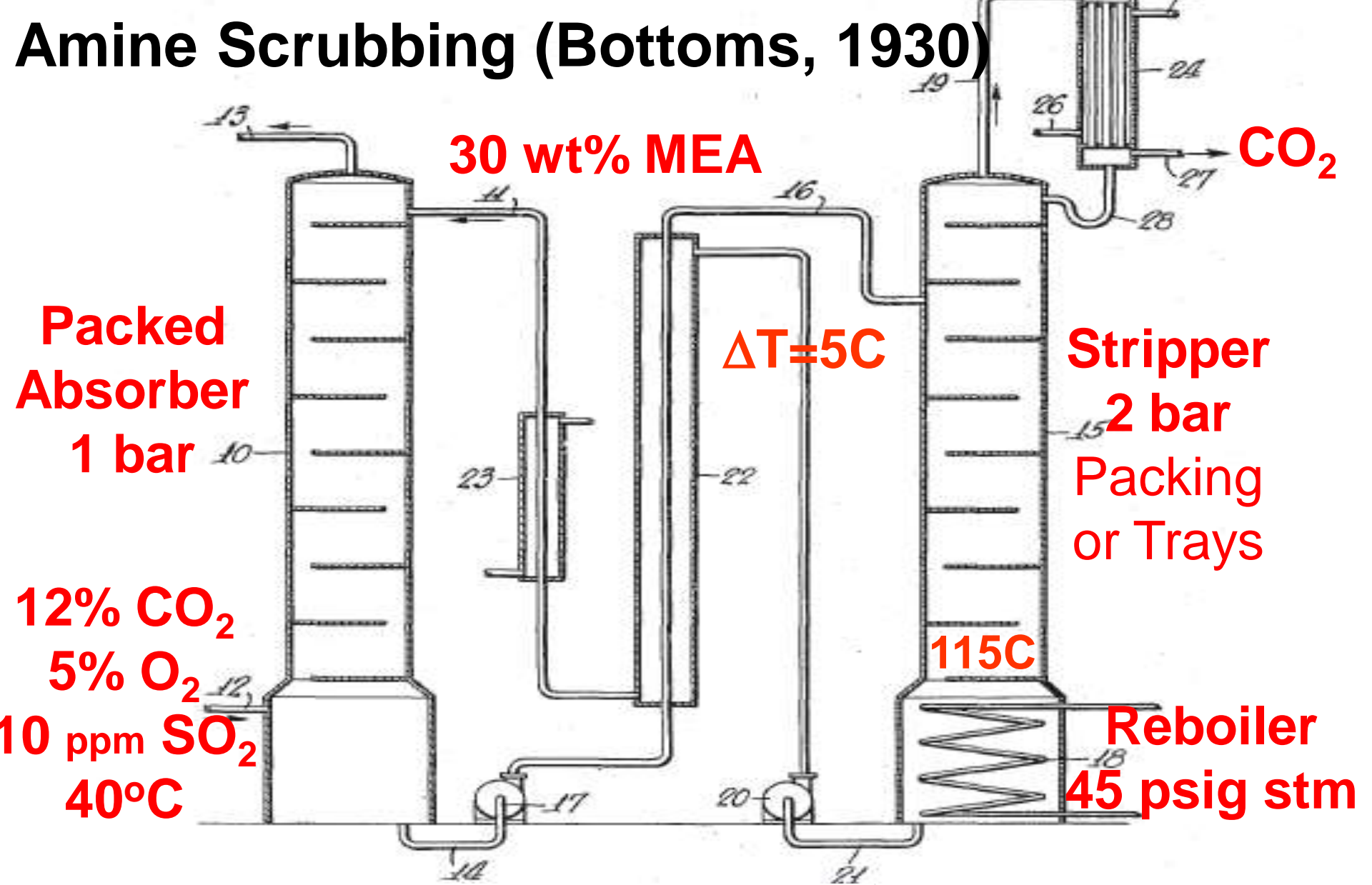
Tail End Technology Ideal for Development, Demonstration, & Deployment

- Low risk
 - Independent, separable, add-on systems
 - Failures impact only Capture and Sequestration
 - Permits on/off operation to address peak load
- Low cost & less calendar time
 - Develop and demonstrate with add-on systems
 - Less demo cost than integrated systems
 - Resolve problems in small pilots with real gas
 - Demo Full-scale absorbers w 100-200 MW gas
 - Ultimately 800 MW absorbers

U.S. Electricity Generation CO₂ Emissions Forecast



Amine Scrubbing (Bottoms, 1930)



Dec. 2, 1930.

R. R. BOTTOMS

1,783,901

PROCESS FOR SEPARATING ACIDIC GASES

Filed Oct. 7, 1930